

VITEEE 2025 Memory Based Question Paper PDF (April 23, 2025) (Shift 1)

1. A body of mass 2 kg is moving with a velocity of 3 m/s. What is its kinetic energy?

- (1) 6 J
 - (2) 9 J
 - (3) 12 J
 - (4) 18 J
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2. A bulb rated 60 W operates for 2 hours. How much energy does it consume in this time?

- (1) 120 J
 - (2) 1200 J
 - (3) 720 J
 - (4) 4320 J
-

3. What is the focal length of a lens if its power is +2 D?

- (1) 0.5 m
 - (2) 1.0 m
 - (3) 2.0 m
 - (4) 0.2 m
-

4. A stone is dropped from a height of 45 m. What is the time taken for the stone to reach the ground?

- (1) 3 s
 - (2) 5 s
 - (3) 6 s
 - (4) 9 s
-

5. A current of 2 A flows through a resistor for 10 minutes. What is the total charge that flows through the resistor?

- (1) 120 C
 - (2) 200 C
 - (3) 240 C
 - (4) 360 C
-

6. A body of mass 10 kg is moving with a speed of 5 m/s. What is the momentum of the body?

- (1) 50 kg m/s
 - (2) 100 kg m/s
 - (3) 200 kg m/s
 - (4) 25 kg m/s
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7. The potential energy of a body at a height of 10 meters is 200 J. What is its mass? (Take $g = 10 \text{ m/s}^2$)

- (1) 2 kg
 - (2) 5 kg
 - (3) 10 kg
 - (4) 20 kg
-

8. What is the resistance of a conductor if the potential difference across it is 12 V and the current flowing through it is 3 A?

- (1) 4 Ω

- (2) $6\ \Omega$
 - (3) $9\ \Omega$
 - (4) $12\ \Omega$
-

9. A convex lens has a focal length of 10 cm. What is the magnification produced when the object is placed 30 cm from the lens?

- (1) 2
 - (2) 3
 - (3) 1.5
 - (4) 4
-

10. What is the molar mass of NaCl?

- (1) 58 g/mol
 - (2) 60 g/mol
 - (3) 62 g/mol
 - (4) 56 g/mol
-

11. Which of the following is an example of a redox reaction?

- (1) $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
 - (2) $\text{NaCl} + \text{AgNO}_3 \rightarrow \text{NaNO}_3 + \text{AgCl}$
 - (3) $\text{C}_2\text{H}_5\text{OH} + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 3\text{H}_2\text{O}$
 - (4) $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
-

12. What is the pH of a solution if the concentration of H^+ ions is $1 \times 10^{-5}\text{ mol/L}$?

- (1) 5
 - (2) 6
 - (3) 4
 - (4) 7
-

13. What is the number of moles in 18 g of water (H_2O)?

- (1) 1 mol
- (2) 0.5 mol

- (3) 2 mol
 - (4) 0.25 mol
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14. Which of the following gases has the highest density at STP?

- (1) O₂
 - (2) N₂
 - (3) CO₂
 - (4) H₂
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15. Which of the following acids is a strong acid?

- (1) H₂SO₄
 - (2) CH₃COOH
 - (3) H₂CO₃
 - (4) HNO₃
-

16. What is the pH of a solution with a H⁺ concentration of 1×10^{-3} mol/L?

- (1) 3
 - (2) 4
 - (3) 5
 - (4) 6
-

17. Which of the following is an example of an exothermic reaction?

- (1) $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$
 - (2) $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
 - (3) $\text{C}_6\text{H}_6 + 15\text{O}_2 \rightarrow 6\text{CO}_2 + 3\text{H}_2\text{O}$
 - (4) $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
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18. What is the molar volume of an ideal gas at standard temperature and pressure (STP)?

- (1) 22.4 L/mol
- (2) 1 L/mol
- (3) 24 L/mol

(4) 12 L/mol

19. Which of the following is a characteristic property of acids?

- (1) They turn blue litmus paper red.
 - (2) They turn red litmus paper blue.
 - (3) They are slippery to touch.
 - (4) They have a bitter taste.
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20. Find the value of x in the equation $2x + 3 = 7x - 8$.

- (1) $x = 1$
 - (2) $x = 2$
 - (3) $x = 3$
 - (4) $x = 4$
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21. The sum of the first 10 terms of an arithmetic progression is 150. If the first term is 10, what is the common difference?

- (1) 5
 - (2) 3
 - (3) 4
 - (4) 2
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22. What is the area of a triangle with base 8 cm and height 6 cm?

- (1) 24 cm^2
 - (2) 30 cm^2
 - (3) 36 cm^2
 - (4) 48 cm^2
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23. Solve for x in the equation $\frac{1}{x} + \frac{1}{x+2} = \frac{5}{6}$.

- (1) $x = 1$
- (2) $x = 2$
- (3) $x = 3$
- (4) $x = 4$

24. Find the solution of the quadratic equation $2x^2 - 3x - 5 = 0$.

- (1) $x = 1, x = -2$
 - (2) $x = -1, x = 2$
 - (3) $x = \frac{5}{2}, x = -1$
 - (4) $x = \frac{-5}{2}, x = 1$
-

25. The ratio of the sum of two numbers to their difference is 5:1. If the sum of the numbers is 18, find the numbers.

- (1) 12, 6
 - (2) 10, 8
 - (3) 9, 9
 - (4) 14, 4
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26. Coding Decoding: If in a certain code language, "APPLE" is written as "BQQMF", how would "MANGO" be written in the same code?

- (1) NBPHP
 - (2) NBQHP
 - (3) NCPHQ
 - (4) MBOHP
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27. Data Interpretation: The following table shows the number of students in five different colleges. Answer the questions based on the data.

College	Number of Students
A	1000
B	1500
C	1200
D	1300
E	900

Question: What is the average number of students in all five colleges?

- (1) 1200
 - (2) 1000
 - (3) 1100
 - (4) 1300
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28. Clock Calendars: If today is Monday, what day of the week will it be after 100 days?

- (1) Thursday
 - (2) Wednesday
 - (3) Tuesday
 - (4) Friday
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29. Simple Arithmetic: What is the value of 15% of 120?

- (1) 12
 - (2) 18
 - (3) 24
 - (4) 15
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30. Poetic Devices: Identify the poetic device used in the following line: "*The wind whispered through the trees.*"

- (1) Simile
 - (2) Personification
 - (3) Alliteration
 - (4) Metaphor
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31. Synonyms/Antonyms: What is the synonym of the word "Eloquent"?

- (1) Simple
 - (2) Unclear
 - (3) Articulate
 - (4) Shy
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32. Idioms/Proverbs: Which of the following idioms means "to do something in a very energetic or enthusiastic manner"?

- (1) To burn the midnight oil
 - (2) To break the ice
 - (3) To go the extra mile
 - (4) To hit the nail on the head
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